Docket No.: 22133-00006-US

Application No. 10/663,835 Amendment dated Reply to Office Action of June 21, 2005

AMENDMENTS TO THE CLAIMS

- (withdrawn and currently amended) A process for the production of a gas diffusion electrode, suitable for use in electrolysis of an aqueous solution of hydrogen chloride, comprising:
 - a) spraying a dispersion comprising a noble metal catalyst and a proton-conducting ionomer in an organic solvent onto an electrically conductive support, said support optionally being provided with a coating comprising an acetylene black/polytetrafluoroethylene mixture, said support comprising woven fabric, braid, net and/or nonwoven material comprising carbon, metal and/or sintered metal, and
 - b) removing the organic solvent.
 - 2. (Withdrawn and currently amended) A diffusion electrode according to claim 13, process according to Claim 1, wherein said catalyst comprises a compound of the formula MeI_xMeII_(6-x)E₈, wherein MeI is molybdenum, MeII is ruthenium, platinum, rhenium, rhodium or palladium, E is sulphur, selenium or chlorine and x is from 0 to 6.
 - 3. (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13</u>, process according to Claim 1, wherein said catalyst comprises a platinum-ruthenium alloy.
 - (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13, said</u>
 process according to Claim 1, further comprising
 - a) spraying a second dispersion comprising proton-conducting ionomer in an organic solvent and
 - b) removing said organic solvent.

Application No. 10/663,835 Amendment dated Reply to Office Action of June 21, 2005 Docket No.: 22133-00006-US

- 5. (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13.</u>

 process according to Claim 1, wherein as a result of said spraying of the dispersion according to a) and removing the solvent according to b), the electrically conductive support is loaded from 0.5 g/m² to 10 g/m², based on the noble metal of the catalyst.
- (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13</u>, process according to Claim 1, wherein said electrically conductive support has a high specific surface area.
- 7. (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13</u>, process-according to Claim 1, wherein said support comprises carbon black.
- 8. (Withdrawn and currently amended) A diffusion electrode process according to Claim 6, wherein the ratio of the mass of the catalyst to the mass of the proton-conducting ionomer is from 1:1 to 15:1.
- (Withdrawn and currently amended) A <u>diffusion electrode process</u> according to claim 8, wherein said ratio is 3:1 to 6:1.
- 10. (Canceled)
- 11. (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13</u>, process according to Claim 1, wherein said dispersion and/or said second dispersion comprises a dispersion of Nafion[®] in alcohol.
- 12. (Withdrawn and currently amended) A <u>diffusion electrode according to claim 13</u>, process according to Claim 1, wherein the dispersion according to steps a) and b) is sprayed on at least two times, and/or the second dispersion according to step c) and d) is sprayed on at least two times.
- 13. (Currently amended) A gas diffusion electrode obtained by a process comprising:

Docket No.: 22133-00006-US

Application No. 10/663,835 Amendment dated Reply to Office Action of June 21, 2005

- a) spraying a dispersion comprising a noble metal catalyst and a protonconducting ionomer in an organic solvent onto an electrically conductive support,
 said support being provided with a coating comprising an acetylene
 black/polytetrafluoroethylene mixture, said support comprising woven fabric,
 braid, net and/or nonwoven material comprising carbon, metal and/or sintered
 metal, and
 - b) removing the organic solvent.

according to Claim 1.

- 14. (Withdrawn) A process for the production of a gas diffusion electrode, suitable for use in electrolysis of an aqueous solution of hydrogen chloride, comprising:
 - a) spraying a dispersion comprising a noble metal catalyst and a protonconducting ionomer in an organic solvent to form a gas diffusion electrode, and
 - b) removing the organic solvent.
- 15. (Currently amended) A gas diffusion electrode <u>suitable for use in electrolysis of an</u>

 <u>aqueous solution of hydrogen chloride</u>, <u>said electrode</u> prepared according to a process

 <u>comprising</u>
 - a) spraying a dispersion comprising a noble metal catalyst and a protonconducting ionomer in an organic solvent on a support comprising woven fabric,
 braid, net and/or nonwoven material comprising carbon, metal and/or sintered
 metal to form a gas diffusion electrode, and
 - b) removing the organic solvent.

of claim 14.

Docket No.: 22133-00006-US

Application No. 10/663,835 Amendment dated Reply to Office Action of June 21, 2005

- 16. (Withdrawn and currently amended) A gas diffusion electrode of claim 15, process according to claim 14 wherein said spraying and removing are conducted at least two times each.
- 17. (Withdrawn and currently amended) A gas diffusion electrode of claim 15, process according to claim-14 wherein after removing said organic solvent, a second dispersion that is the same or different than said dispersion is sprayed onto said support.
- 18. (canceled)
- 19. (currently amended) A gas diffusion electrode that has been prepared by spraying and drying catalyst on a support, said support comprising woven fabric, braid, net and/or nonwoven material comprising carbon, metal and/or sintered metal, and and wherein sintering is not employed in the preparation thereof.
- (Original) A gas diffusion electrode according to claim 19, wherein said electrode is loaded with from 0.5 to 10g/m² of said catalyst.
- (Original) A gas diffusion electrode according to claim 19, wherein said catalyst is a noble metal catalyst.